

Department of Physics, University of Maryland
College Park, MD 20742-4111

Physics 603

HOMEWORK ASSIGNMENT #1

Spring 2014

Due date for problems on Thursday, Feb. 6 [deadline on Feb. 11].

1. a) Find the Legendre transform $\Psi(P)$ of $\Gamma(X) = X^3$.

b) Carry out an explicit Legendre transformation for a more complicated function: Consider the thermodynamic potential Y :

$$Y(U,X) = A + BU + CX^2 + DU^2 + EU^2X^2$$

Calculate $W = (\partial Y / \partial U)_X$ and $P = (\partial Y / \partial X)_U$

Construct explicitly the thermodynamic potential $\Psi(W,X)$ and from it verify the relations

$$U = -(\partial \Psi / \partial W)_X \text{ and } P = (\partial \Psi / \partial X)_W$$

2. a) Derive the Maxwell relation associated with $H(S,p)$.

b) Verify the Maxwell relation $(\partial S / \partial V)_T = (\partial p / \partial T)_V$. From which thermodynamic function does it originate?

c) Extend U to $U(S,V,m)$ and G to $G(T,p,m)$ by adding $H dm$ (i.e. the magnetic work ON an object is $H dm$, analogous to $-p dV$ for mechanical work), and write down the new Maxwell relations involving H and/or m that result. (Note that this formulation includes the magnetic energy of the object; cf. the Kittel posting. Here m is taken as extensive, because M curiously is conventionally reserved for magnetization density (i.e., magnetization per volume). In older books, the extensive magnetization is often written \mathfrak{M} [actually in Germanic Faktur, but Old English is the closest on my computer].)

3. (essentially Kardar 1-7) For an elastic filament it is found that, at a finite range in temperature, a displacement x requires a force

$$J = ax - bT + cTx ,$$

where a , b , and c are constants. Furthermore, its heat capacity at constant displacement is proportional to temperature: $C_x [= T \partial S / \partial T]_x = A(x) T$.

a) Use an appropriate Maxwell relation to calculate $\partial S / \partial x|_T$.

b) Show that A must be independent of x : i.e., $dA/dx = 0$.

c) Calculate $S(T,x)$, assuming $S(0,0) = S_0$.

d) Show that the heat capacity at constant tension, $C_J = T \partial S / \partial T|_J$, can be written as

$$C_J = T \left[A + \frac{(ab - cJ)^2}{(a + cT)^3} \right]$$

4.

Problem I.3

Equilibrium is a central concept in Statistical Physics.

- (a) ~~[6 points]~~ Give a definition of equilibrium for
 - (a) an isolated system
 - (b) a pair of interacting systems
- (b) ~~[6 points]~~
 - (a) For a closed isolated system, give two examples of experimentally measurable thermodynamic variables, their definitions, and what equilibrium implies about them.
 - (b) For a system that can exchange particles, give one more example of an experimentally measurable thermodynamic variable and its definition.
- (c) ~~[7 points]~~ A colleague gives you a crystalline solid sample and asks you to examine its electronic properties (e.g. voltage V versus current I).
 - (a) What are the thermodynamic variables and potentials that are relevant to your measurement?
 - (b) As an experimentalist, why should you concern yourself with the sample equilibrium?
- (d) ~~[6 points]~~ A laboratory is never in equilibrium, in the sense of an isolated system. Give four examples of likely nonequilibrium conditions for a scientific laboratory in this building (or another campus building if you prefer).