

## Solution to problems 4 & 5 Exam II

4 a. The Energy  $E$  is given by

$$E_n = \frac{n^2 h^2}{8mL^2} \quad n = 1, 2, \dots$$

Here,  $L = 0.5 \text{ nm} = 0.5 \times 10^{-9} \text{ meters}$

$$m = 9.11 \times 10^{-31} \text{ Kg}$$

$$h = 6.63 \times 10^{-34} \text{ Js}$$

Plugging values

$$E_n = n^2 \times 2.4 \times 10^{-19} \text{ J}$$

The first two levels are for  $n=1$  and  $n=2$

$$E_{n=1} = 2.4 \times 10^{-19} \text{ J}$$

$$E_{n=2} = 9.6 \times 10^{-19} \text{ J}$$

b. The energy lost must be the difference between  $E_{n=2}$  and  $E_{n=1}$

$$\Rightarrow \Delta E = E_2 - E_1 = 7.2 \times 10^{-19} \text{ J}$$

c. The energy lost, calculated in b, must now be the energy of the electron (assuming no energy is lost)

$$\Rightarrow E_{\text{photon}} = \Delta E = 7.2 \times 10^{-19} \text{ J}$$

We now recall that  $E_{\text{photon}} = \frac{hc}{\lambda}$

$$\Rightarrow \frac{hc}{\lambda} = 7.2 \times 10^{-19}$$

$$\Rightarrow \lambda = \frac{h \cdot c}{7.2 \times 10^{-19}} = \frac{6.63 \times 10^{-34} \times 3 \times 10^8}{7.2 \times 10^{-19}}$$

$$= 2.76 \times 10^{-7} \text{ m}$$

$$= 276 \times 10^{-9} \text{ m} = 276 \text{ nm}$$

5. The width is given by

$$W = \frac{2\lambda L}{a}$$

here  $a = 1 \mu\text{m} = 10^{-6} \text{ m}$ ;  $L = 2 \text{ m}$

To calculate  $\lambda$ , we use the de-Broglie relation

$$\lambda = \frac{h}{p} = \frac{h}{mv}$$

$$\Rightarrow W = \frac{2 \times \frac{h}{mv} \times L}{a} = \frac{2hL}{mva}$$

$$= \frac{2 \times 6.6 \times 10^{-34} \times 2}{9.1 \times 10^{-31} \times 4 \times 10^6 \times 10^{-6}}$$

$$= 0.73 \times 10^{-3} \text{ m}$$

$$= 0.73 \text{ mm}$$