

# Physics 117

## Homework Solutions to HW Set # 8

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| By | <u>QH</u> |
| On | / /       |

CH 10: CQ 23, 33, 34; Ex 7, 13, 17

CH 11: CQ 5, 9; Ex 3, 8

Ch 10 CQ 23. To an observer in the lab frame, the time interval measured by a clock (or the average lifetime of a decaying particle) moving through the lab at speed,  $v$ , is dilated (i.e., increased) by the adjustment factor,  $\gamma = (1 - v^2/c^2)^{-1/2} \geq 1$ , as compared with a clock at rest in the lab.

Then if the mu meson lifetime at rest (i.e. observed in its own rest frame) is  $T_\mu^0$ , the lab observer will measure this lifetime to be  $T_\mu^\nu = \gamma T_\mu^0 \geq T_\mu^0$ . Thus the lab lifetime of 8  $\mu s$  ( $\mu s = \text{micro-seconds}$ , and this  $\mu$ , - for "micro" =  $10^{-6}$  has nothing to do with the nucleon symbol,  $\mu$ , for the mu-meson, or muon.) is greater than that of the muon at rest by the factor  $\gamma \geq 1$ , and  $T_\mu^0 \leq T_\mu^\nu$ . I.e. muon's lifetime at rest is less than its lifetime measured in a lab through which it moves.

Ch 10 CQ 33. If  $T_\mu^0 = 2.2 \mu s$  is the average lifetime of a muon at rest (i.e. as measured by an observer in its own rest frame), then at a speed  $c$  (and no speed can be greater!), it could travel  $c \cdot T_\mu^0 = 3 \times 10^8 \text{ m/s} \cdot 2.2 \times 10^{-6} \text{ s} = 660 \text{ m}$ , at most, before decaying.

But if the muon is moving in a lab frame on earth, then its lifetime is dilated by a factor of  $\gamma$ , and the maximum distance would be  $c \cdot \gamma T_\mu^0 >> c T_\mu^0 = 660 \text{ m}$ .

In this case the observer must be in the reference system through which the muon is moving (e.g. earth's lab frame).

An observer in the muon rest frame would report (a) that muon is at rest, and earth is moving with velocity  $-\vec{v}$ , just opposite to what earth's observer says about muon, and

(b) That the distance from upper atmosphere to earth's surface, which earth/lab measures to be  $D$  is actually "Contracted" to the smaller distance  $D/\gamma = D'$  for the observer at rest on the muon. He says that the small earth-atmosphere distance  $D'$  travels past the muon at speed

$$\frac{D'}{T_\mu^0} = v_E = \frac{D}{\gamma T_\mu^0}, \text{ since } D' = \frac{D}{\gamma} \text{ is a contracted moving length.}$$

The earth lab observer says that the muon is moving

with speed  $v_\mu = D/T_\mu^0 = D/\gamma T_\mu^0$ , given in his frame by the distance  $D$  (at rest & uncontracted) divided by the dilated lifetime of the moving meson  $T_\mu^0 = T_\mu^0 \gamma$ .

The two observers agree only on the value of  $|v|$ .

Ch 10 CQ 34. If the muon's arrival at earth's surface after traversing  $D \gg 660\text{ m}$  is to be explained in terms of length contraction, then the distance travelled must be moving in the frame of the observer. If  $D$  is the height of the top of the earth's atmosphere <sup>(in the earth's rest frame)</sup> and is to be moving, then the observer is in the rest frame of the moving muon. He measures objects moving in his own rest frame to be shortened (contracted) in the direction parallel to their velocity by a factor,  $1/\gamma$ . Then he measures the height  $D$  to be the much smaller value  $D' = D/\gamma \ll 660\text{ m}$ , and says that it can pass by the muon in the span of its lifetime in its rest frame,  $T_\mu^0$ .

To the earthlab observer  $D$  is at rest, so no question of contraction arises. On the other hand the muon is moving, so its internal clock (which defines its average lifetime) shows a time dilation. He explains the great distance travelled ( $D$ ) by his observation that the muon's average lifetime is  $\gamma T_\mu^0 = T_\mu^{av} \gg T_\mu^0$ .

Thus the two observers disagree both about clock times & distances ... they agree only on the relation speed  $v = \frac{D'}{\gamma T_\mu^0} = \frac{D}{T_\mu^{av}} = \frac{D}{T_\mu^0}$ .

Ch 10 Ex 7.

$$\gamma_T^{\sigma} = \gamma T_{\pi}^0 = (1 - v^2/c^2)^{-1/2} T_{\pi}^0$$

$$2.69 \times 10^{-9} s = \frac{1}{1 - (0.99)^2} \gamma^{-1/2} T_{\pi}^0$$

$$\frac{2.69 \times 10^{-9}}{7.09} = T_{\pi}^0 = 3.79 \times 10^{-10} \text{ sec}$$

= Lifetime of pion at rest

Ex 13 Exactly the same answer since she is observing the train in the rest frame of the train. An observer

in the station's rest frame, however, would measure the moving train's length to be contracted by a factor  $1/\gamma$  ( $\leq 1$ ). Then she again measures 200 m

Ex 17. Since  $p^R = \gamma m v$  is the relativistic momentum (replacing  $p^{NR} = m v$ , non-relativistically) we have  $F_R \Delta t = \text{impulse} = \text{change in momentum} = p_f^R - p_i^R$ .

and  $F_R \Delta t = \gamma_f m v_f - 1 \cdot m v_0 = \gamma_f \cdot m v_f$ .

The corresponding non-relativistic result is  $F_{NR} \Delta t = 1 \cdot m v_f$ .

Since  $\Delta t$  is the same for both  $F_R = \frac{\gamma_f \cdot m v_f}{\Delta t} = \gamma_f \cdot F_{NR}$

The relativistic force must be larger by the factor  $\gamma$ :  $\gamma = [1 - (1 - 10^{-4})^2]^{-1/2} = (1 - 1 + 2 \cdot 10^{-4} - 10^{-8})^{-1/2} = \sqrt{1/2 \times 10^{-4}}$

$$= \frac{10^2}{\sqrt{2}} = 70.7$$

$$\text{Therefore, } F_R = (9.5)(70.7) = 672 N$$

Note that calculator rounding error can lead to false results when the answer gets too small. Then writing  $\frac{1}{1-\epsilon} = 1 + \epsilon$  yields  $(1 - v^2/c^2)^{-1/2} = [1 - (1-\epsilon)^2]^{-1/2} = [1 - (1 - 2\epsilon + \epsilon^2)]^{-1/2} \approx \frac{1}{\sqrt{2}\epsilon}$  for small  $\epsilon$ . Since  $\epsilon^2 \ll \epsilon$  when  $\epsilon \ll 1$ .

Ch 11: CQ 5, 9 / Ex 3, 8.

Q5) Additional examples in which experimental results agree with a model strengthen our belief in it. However a model can never be proven true

II: Q9) Salt granite and water <sup>are</sup> not elements. Each is composed of more than one element

II: Ex 3) Only 4 g of hydrogen is required to use up 32 g of oxygen fully. Amount of water that can be formed =  $4 \text{ g} + 32 \text{ g} = [36 \text{ g}]$ , and 4 g of H will be left over

II: Ex 8) Molecular mass of sulphur 2 amu  
1 mole of sulfur weighs 32 g

There are Avogadro number of molecules  
one mole  $N_A = 6 \times 10^{23}$

32 g of sulfur contain  $N_A = 0.2 \times 10^{23}$  molecules

g of sulfur contains  $\frac{6.02 \times 10^{23}}{(32)} \text{ molecules}$

$$= 1.88 \times 10^{22} \text{ molecules}$$